



## The synthesis of phthalazin-1(2*H*)-ones and 3-hydroxyisobenzofuran-1(3*H*)-ones via the ring contraction of tropones

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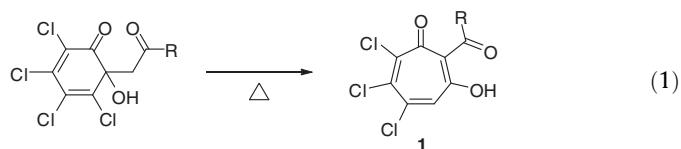
### ABSTRACT

Phthalazin-1(2*H*)-ones and 3-hydroxyisobenzofuran-1(3*H*)-ones were synthesized by the reactions of tropones with hydrazines and alcohols, respectively, via the ring contraction.

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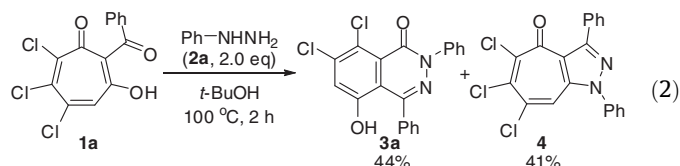
Cycloheptatrienone skeleton (troponone) is the key structural element in the troponoid family, which presents the interesting non-benzenoid aromatic structure.<sup>1</sup> Troponone derivatives are not only found in a wide range of biologically active natural products but are also useful building blocks in synthetic chemistry.<sup>2</sup> Therefore, the synthesis, structure, and reactivity of the troponoids have attracted considerable interest to organic chemists. Rather useful reactions of troponone nucleus are cycloaddition<sup>3</sup> and annulations,<sup>4</sup> providing accesses to functionalized bi- or multi-cyclic compounds. On the other hand, base-induced rearrangement of tropones to benzenoid compounds is also a well-known reaction pathway for troponone derivatives.<sup>2</sup> However, the formation of functionalized bicyclic products by the ring contraction of tropones has been virtually untouched.<sup>5</sup>

Recently, we developed an efficient ring expansion method for the synthesis of multiple functional troponone derivatives **1** (Eq. 1).<sup>6</sup> As the troponone **1** has three chloro-substituents, one active hydroxyl group and  $\beta$ -diketone structure, several interesting reactions were expected. Herein, we report our preliminary studies on the ring contraction of **1** introduced by hydrazines and alcohols.



Due to  $\beta$ -dicarbonyl structure of **1**, hydrazines **2** were initially used to evaluate the reactivity of **1**. Phthalazin-1(2*H*)-one **3a** was obtained unexpectedly together with an annulation product, pyrazol **4** (Eq. 2). Phthalazinones are biologically active compounds<sup>7</sup> and have attracted diverse areas of interests.<sup>8</sup> Several approaches have been reported for the synthesis of phthalazinones.<sup>9</sup> We

envisioned that the present transformation could provide an alternative strategy to synthesize phthalazin-1(2*H*)-ones if the ring contraction occurred selectively.



A variety of hydrazines were tested to realize our hypothesis. Interestingly, the electron effect of hydrazines dramatically influenced the selectivity of the ring contraction of **1a**. Aromatic hydrazines **2** with electron-withdrawing group selectively afforded the ring-contraction products **3** (Table 1). 2,4-Dinitrophenyl hydrazine **2b** reacted with **1a** and gave the corresponding product **3b** in 94% yield (entry 1). Perfluorophenyl hydrazine **2c** also led to a good yield of the desired product **3c** (entry 2). A moderate yield of **3d** was obtained due to the formation of a **4**-type cycloaddition product in ca.6% yield (entry 3). Methyl-substituted troponone **1b** also selectively transformed into the corresponding ring-contraction products (entries 4 and 5). Although furan-substituted phthalazin-1(2*H*)-one **3g** was formed exclusively under the standard conditions, a slightly lower yield was attributed to a lower conversion of **1c** (entry 6).

A tentative mechanism for the formation of **3** is proposed (Scheme 1). The addition of **2** to **1** gives **A**, followed by the elimination of H<sub>2</sub>O to afford **B**. We hypothesized that hydrazines **2** with an electron-withdrawing group would make this step feasible. Subsequently, intramolecular addition led to a five-membered intermediate **C**, followed by an electrocyclic reaction to give a tricyclic intermediate **D**. Finally, the three-membered ring opening generates the product **3**.

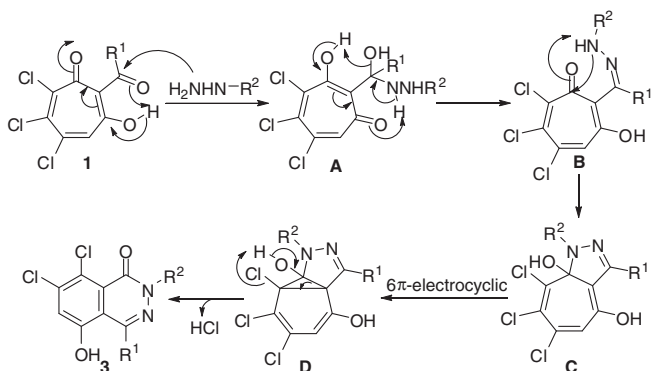
Interestingly, various alcohol **5** also induced the ring contraction of tropones **1b** to give 3-hydroxyisobenzofuran-1(3*H*)-ones **6** (Table 2). A 95% yield of **6a** was obtained when methanol **5a** was used as solvent (entry 1). The decreasing yields were observed when

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**Table 1**  
The reactions of tropones **1** with hydrazines **2**<sup>a</sup>

Entry	<b>1</b>	<b>2</b>	<b>3</b>	Isolated yields (%)
1				<b>3b</b> 94
2	<b>1a</b>			<b>3c</b> 89
3	<b>1a</b>			<b>3d</b> 74
4		<b>2b</b>		<b>3e</b> 81
5	<b>1b</b>	<b>2c</b>		<b>3f</b> 86
6		<b>2c</b>		<b>3g</b> 68

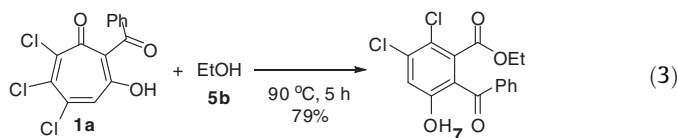
<sup>a</sup> Conditions: **1** (0.1 mmol), **2** (0.3 mmol); under nitrogen atmosphere.



**Scheme 1.** A tentative mechanism for the formation of **3**

ethanol **5b** and *iso*-propanol **5c** were applied (entries 2 and 3). *tert*-Butyl alcohol **5d** does not induce the ring contraction (entry 4). Moreover, other alcohols **5e** and **5f** gave moderate yields of the corresponding products **6e** and **6f** (entries 5 and 6). Related to the hydrazine-induced ring contraction, we rationalized that the hemiketal intermediate **E** is generated firstly (Scheme 2). Keto-enol tautomerization gives the intermediate **F**. Oxygen instead of nitrogen attacks the carbonyl group of the troponone skeleton to generate a four-membered intermediate **G**, followed by the electrocyclic reaction and the three-membered ring opening to give the product **6**.

It should be noted that phenol derivative **7** instead of 3-hydroxyisobenzofuran-1(3*H*)-ones **6** was obtained when **1a** reacted with ethanol (Eq. 3). We hypothesized that a **6**-type intermediate was formed firstly in this reaction, followed by transesterification to give the ring-opening product **7**.



**Table 2**  
The reactions of tropones **1b** with alcohols **5**<sup>a</sup>

Entry	<b>6</b>	<b>7</b>	Isolated yield (%)
1 <sup>b</sup>	MeOH <b>5a</b>		<b>6a</b> 95
2 <sup>c</sup>	EtOH <b>5b</b>		<b>6b</b> 85
3			<b>6c</b> 70
4		–	<sup>d</sup>
5			<b>6e</b> 68
6			<b>6f</b> 54

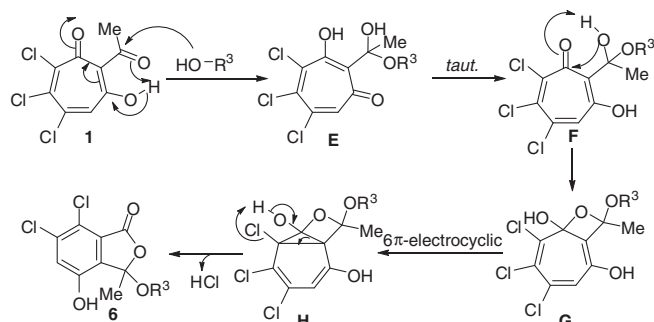
<sup>a</sup> Conditions: **1** (0.1 mmol), **5** (1.0 mL); under nitrogen atmosphere.

<sup>b</sup> 80 °C.

<sup>c</sup> 90 °C.

<sup>d</sup> No reaction.

In summary, we have reported a novel hydrazine-/alcohol-induced ring contraction of tropones, which provides a new protocol



**Scheme 2.** A tentative mechanism for the formation of **6**

for the synthesis of the functionalized phthalazin-1(2*H*)-ones and 3-hydroxyisobenzofuran-1(3*H*)-ones. Further investigation of the reactivity of the multiple functional tropones **1** and the applications of phthalazinones **3** and isobenzofuranones **6** are in progress.

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### Supplementary data

Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.tetlet.2010.08.017](https://doi.org/10.1016/j.tetlet.2010.08.017).

### References and notes

- Dewar, M. J. S. *Nature* **1945**, *155*, 50.
- (a) Morita, N.; Toyota, K.; Ito, S. *Heterocycles* **2009**, *78*, 1917; (b) Pietra, F. *Chem. Rev.* **1973**, *73*, 293; (c) Pietra, F. *Acc. Chem. Res.* **1979**, *12*, 132; (d) Pauson, P. L. *Chem. Rev.* **1955**, *55*, 9.
- (a) Li, P.; Yamamoto, H. *J. Am. Chem. Soc.* **2009**, *131*, 16628; (b) Trost, B. M.; McDougall, P. T. *Org. Lett.* **2009**, *11*, 3782; (c) Nair, V.; Poonoth, M.; Vellalath, S.; Suresh, E.; Thirumalai, R. *J. Org. Chem.* **2006**, *71*, 8964; (d) Du, Y.; Feng, J.; Lu, X. *Org. Lett.* **2005**, *7*, 1987.
- (a) Gao, W.; Sun, M.; Li, Y.; Li, W.; Imafuku, K. *J. Heterocycl. Chem.* **2009**, *46*, 1302; (b) Ishizu, T.; Mori, M.; Kanematsu, K. *J. Org. Chem.* **1981**, *46*, 526.
- (a) Kogler, H.; Fehlhaber, H.-W.; Leube, K.; Durckheimer, W. *Chem. Ber.* **1989**, *122*, 2205; (b) Schenck, G. O.; Brahler, B.; Cziesla, M. *Angew. Chem.* **1956**, *68*, 247.
- Li, H.; Li, W.; Li, Z. *Chem. Commun.* **2009**, 3264.
- (a) Jain, R. P.; Vederas, J. C. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3655; (b) Al-Assar, F.; Zelenin, K. N.; Lesiovskaya, E. E.; Bezant, I. P.; Chakchir, B. A. *Pharm. Chem. J.* **2002**, *36*, 598.
- (a) Pakulska1, W.; Malinowski, Z.; Szczesniak, A. K.; Czarnecka1, E.; Epsztajn, J. *Arch. Pharm. Chem. Life Sci.* **2009**, *342*, 41; (b) Li, Y.-X.; Luo, Y.-P.; Xi, Z.; Niu, C.; He, Y.-Z.; Yang, G.-F. *J. Agric. Food Chem.* **2006**, *54*, 9135.
- (a) Vina, D.; del Olmo, E.; Lopez-Perez, J. L.; Feliciano, A. S. *Tetrahedron* **2009**, *65*, 1574; (b) Grasso, S.; De Sarro, G.; De Sarro, A.; Micale, N.; Zappala, M.; Puia, G.; Baraldi, M.; De Micheli, C. *J. Med. Chem.* **2000**, *43*, 2851; (c) Attanasi, O. A.; Filippone, P.; Fiorucci, C.; Mantellini, F. *Tetrahedron Lett.* **1999**, *40*, 3891; (d) Lefkaditis, D. A.; Nicolaidis, D. N.; Papageorgiou, G. K.; Stephanidou-Stephana, J. *J. Heterocycl. Chem.* **1990**, *27*, 227.